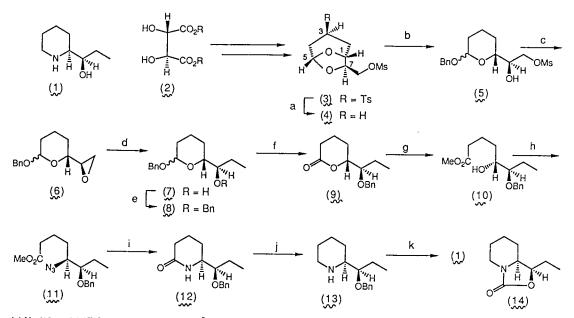
## ENANTIOSPECIFIC SYNTHESIS OF OPTICALLY ACTIVE NATURAL (+)-CONHYDRINE FROM (S,S)-TARTARIC ACID

Yukio Masaki,\* Toshihiro Imaeda, Kinnosuke Nagata, Hirohisa Oda, and Akichika Ito Gifu Pharmaceutical University, 5-6-1 Mitahora-Higashi, Gifu 502, Japan

Summary: The first enantiospecific synthesis of (+)-conhydrine, one of the poisonous alkaloids of the hemlock was achieved via partial ring opening of 6,8-dioxabicyclo[3.2.1]-octane skeleton prepared from (S,S)-tartaric acid.

(+)-Conhydrine  $(1)^{1)}$  is one of the poisonous alkaloids of the hemlock, <u>Conium</u> maculatum, whose extracts were used in the ancient Greece for the execution of criminals. The structure of (+)-conhydrine including the absolute stereochemistry has been established as (2S)-((1'R)hydroxypropyl)piperidine (1).<sup>2)</sup> Although syntheses of optically active natural 1 by way of optical resolution of racemic 1 and of a synthetic intermediate 2-pyridylethylcarbinol have been reported,  $3^{(1)}$  there has been no enantiospecific synthesis of  $1^{(4)}$  starting with a configurationally defined chiral compound, that means synthetic verification of the structure of natural (+)-1. We wish to report here the first enantiospecific synthesis of natural (+)-1. from (S,S)-tartaric acid (2 R=H). The synthesis of (+)-1 described is featured by mild alcoholysis of 6,8-dioxabicyclo[3.2.1]octane skeleton prepared from 2 (R=H) and introduction of the nitrogen functionality with inversion of configuration of one of the hydroxyl groups. (+)-(1R,5R,7R)-7-Mesyloxymethyl-6,8-dioxabicyclo[3.2.1]octane (4) ([] 20 +62.7° (CHCl<sub>3</sub>)) was synthesized as an oil by reductive desulfurization of the 3-tosyl-derivative (3) (mp. 106-108 °C,  $[\alpha]_D^{20}$  +49.6° (CHCl<sub>3</sub>)) easily obtained in 54% overall yield from diethyl (S,S)tartrate (2, R=Et) according to the reported method.<sup>5)</sup> Although it has been known that unsubstituted 6,8-dioxabicyclo[3.2.1]octane is highly susceptible to polymerize in the presence of a Lewis acid such as  $BF_3$ -Et<sub>2</sub>0,<sup>6)</sup> we were very curious to open the 7-substituted bicyclic (4) by alcoholysis producing a pyranoid acetal of type (5). The desired partial ring opening was realized by treatment of 4 with benzyl alcohol in the presence of  $BF_3$ -Et $_20$  at the room temperature to lead to an acetal (5) in good yield (74%). Homologation with the Gilman's reagent on the terminal position of the epoxide ( $\underline{6}$ ) obtained by treatment of  $\underline{5}$  with  $K_2CO_2$ , gave an alcohol (7). Protection of the alcohol as benzyl ether (8) followed by oxidation of the acetal portion on treatment successively with m-chloroperbenzoic acid (MCPBA) and  $Et_2N^{5b}$ 

afforded a lactone (9) (IR  $V_{C=0}$ :1730 cm<sup>-1</sup>). Methanolysis of 9 followed by the Mitsunobu type reaction of the ester-alcohol (10) using hydrazoic acid<sup>7</sup>) gave an azido ester (11) ([M]<sub>D</sub><sup>20</sup> +7.7° (CHCl<sub>3</sub>); IR  $V_{C=0}$  1730,  $V_{N_3}$  2090 cm<sup>-1</sup>). Cyclization of the intermediate amino ester obtained by reduction of 11 led to a lactam (12) (M]<sub>D</sub><sup>20</sup> +9.4° (CHCl<sub>3</sub>); IR  $V_{C=0}$  1645 cm<sup>-1</sup>) in 82% overall yield. Reduction of the lactam (12) with LiAlH<sub>4</sub> afforded O-benzylconhydrine (13) which was deprotected by hydrogenolysis to lead in 70% overall yield to (+)-conhydrine (1) as a crystal (mp. 119-121 °C;  $[M]_D^{20} +8.9°$  (EtOH)) (1it.<sup>1a)</sup> mp. 121 °C;  $[M]_D^{18} +10.0°$  (EtOH)). The <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, and IR spectra were identical with those for the authentic racemic 1.



(a) Na (10 equiv) / EtOH (10 equiv) / THF / -20°C / 2h (78%); (b) PhCH2OH / BF3 Et2O (2eq) / r.t. / 15h (74%) (c) K<sub>2</sub>CO<sub>3</sub> / MeOH / 0°C / 0.5h (92%); (d) Me<sub>2</sub>CuLi / Et<sub>2</sub>O / -5°C / 4h (90%); (e) NaH / PhCH<sub>2</sub>Br / DME / r.t. / 15h (79%) (f) i) MCPBA/BF3-Et2O(cat.)/CH2Cl2/r.t./2h ii) Et3N/0°C/1.5h (75%);(g) K2OO3/MeOH/0°C/2h (quant.) (h) HN<sub>3</sub> / Ph<sub>3</sub>P / DEAD / benzene / r.t. / 2h (75%); (i) i) H<sub>2</sub> / Pd-black / MeOH ii) toluene / 120°C / 15h (82%) (j) LiAIH<sub>4</sub> / THF / r.t. / 0.5h (80 %); (k) H<sub>2</sub> / 5% Pd-C / EtOH / concd.HCl (cat.) (88 %)

The synthetic (+)-1 was converted by a known method  $^{4b)}$  into a oxazolidone derivative (14) which showed also identical spectral properties with those for the authentic racemic one  $((\pm)-14)^{4b}$  and which did not show any trace of the signals corresponding to the three form  $((\pm)$ -cpi-14)<sup>8)</sup> in the <sup>1</sup>H-NMR spectrum.

Acknowlcdgements: We thank Professor M. Vaultier, Universite de Rennes, for providing us with <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra of  $(\pm)$ -1,<sup>4</sup>C) and also are indebted to Professor T. Shono, Kyoto University, and Professor S. Kano and Dr. S. Shibuya, Tokyo College of Pharmacy, for communicating us <sup>1</sup>H-NMR and IR spectra of  $(\pm)$ -1, <sup>4b</sup>  $(\pm)$ -epi-1, <sup>8</sup>  $(\pm)$ -14, <sup>4b</sup> and  $(\pm)$ -epi-14<sup>8</sup> for identification and comparison.

## References and Notes

- a) E. Späth and E. Adler, <u>Monatsh. Chem.</u>, <u>63</u>, 127 (1933) and references cited therein;
   b) Recently, conhydrine (<u>1</u>) (natural erythro form) sometimes has been designated α-conhydrine and the corresponding three form (epi-1) β-one: see ref. 4c), 8) and V. Rato-

- conhydrine and the corresponding threo form (epi-1) \$\mathcal{B}\$-onc: see ref. 4c), 8) and V. Ratovelomanana, J. Royer, and H-P. Husson, <u>Tetrahedron Lett.</u>, 26, 3803 (1985).
  2) R.K. Hill, <u>J. Am. Chem. Soc.</u>, 80, 1609 (1958) and references cited therein.
  3) a) F. Galinovsky and H. Mulley, <u>Monatsh. Chem.</u>, 79, 426 (1948); b) G. Fodor and E. Bauerschmidt, <u>J. Heterocycl. Chem.</u>, 5, 205 (1968).
  4) For syntheses of racemic conhydrine see: a) G. Stork, R.M. Jacobson, and R. Levitz, <u>Tetrahedron Lett.</u>, 20, 771 (1979); b) T. Shono, Y. Matsumura, and T. Kanazawa, <u>ibid.</u>, 24, 4577 (1983); c) S. Pilard and M. Vaultier, <u>ibid.</u>, 25, 1555 (1984).
  5) a) Y. Masaki, Y. Serizawa, K. Nagata, H. Oda, H. Nagashima, and K. Kaji, <u>Tetrahedron Lett.</u>, 27, 231 (1986) and references cited therein; b) Y. Masaki, K. Nagata, and K. Kaji, <u>Chem. Lett.</u>, 1983, 1835.
  6) K. Kobayashi, H. Sumitomo, H. Ichikawa, and H. Sugiura, <u>Polymer J.</u>, 18, 927 (1986) and references cited therein.
- references cited therein.
- 7) H. Loibner and E. Zbiral, <u>Helv. Chim. Acta</u>, <u>59</u>, 2100 (1976).
  8) S. Kano, T. Yokomatsu, Y. Yuasa, and S. Shibuya, <u>Hetereocycles</u>, <u>24</u>, 621 (1986).

(Received in Japan 18 July 1989)